

Gigantic Reduction of Energy Consumption of Molecule Sensors by Pulse-heating of Suspended SnO₂ Nanowire Device

Gang Meng, Kazuki Nagashima, Fuwei Zhuge, Masaki Kanai, Yong He and Takeshi Yanagida

Institute for Materials Chemistry and Engineering, Kyushu University

Integrating sensors for molecule species into portable electronic devices is strongly desired for forthcoming sensor network society. However, high energy consumption ($\sim$ mJ) and high temperature ($\sim$ 300°C) required for operating conventional gas sensors have been bottleneck issues to apply these for CMOS electronics and also emerging wearable electronics. Here we demonstrate a rational sensing methodology, which substantially reduces the energy consumption of gas sensors down to $\sim$ pJ/sec. Our methodology utilizes (1) a pulse self-Joule heating of suspended SnO₂ nanowire device and (2) the short thermal relaxation time down to microseconds. These features allow us to sense volatile molecules at the energy consumption of $\sim$ pJ/sec via heating up only the local sensing part within a short time by applying a pulse voltage. We show the feasibility of the present methodology for sensing NO₂ (100ppb) by applying a pulse voltage down to microseconds. Surprisingly, the sensitivity can be significantly enhanced by utilizing the present pulse method when compared with conventional continuous heating method. Furthermore, we successfully demonstrate the applicability of the present methodology for volatile molecule sensors on flexible PEN substrate.

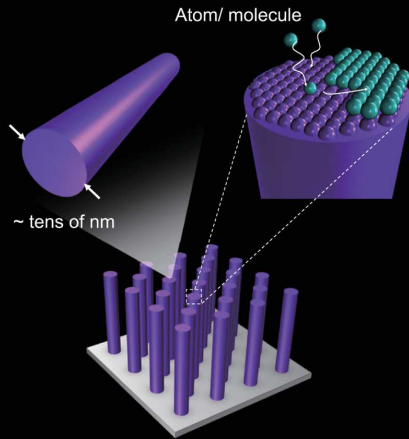
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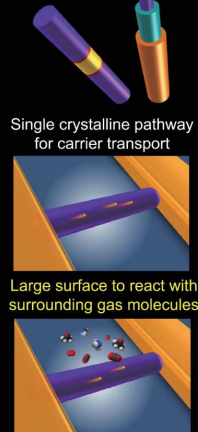
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Introduction

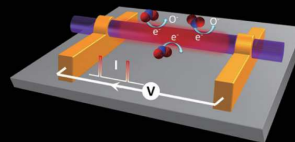
Single-crystalline bottom-up nanowires



1D heterostructures



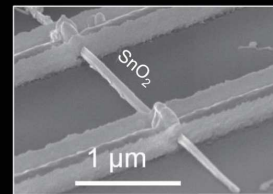
Opportunities offered by a suspended nanowire



► Small Joule power to heat it up
 $\Delta Q = c \cdot m \cdot \Delta T$
 ► Short thermal relaxation time
 $\tau_{\text{relaxation}} \propto \frac{C_P \cdot l_0^2}{\kappa}$

Objective:

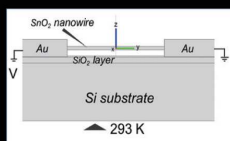
- 1) Evaluate heat transport in self-heated nanowire
- 2) Reduce energy consumption of sensor (< nJ/s)
- 3) Fully explore advantages of suspended nanowire device



Result & discussions

Evaluate Joule heat in nanowire

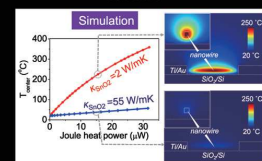
- Simulation model -



Specific heat capacity Thermal conductivity
 Density $\rho = \frac{1}{V} = \frac{1}{\pi \cdot d^2 \cdot L}$
 Heat sink (Joule heat)

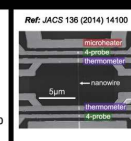
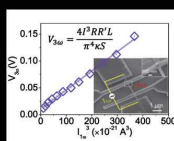
Thermal conductivity κ maybe crucial

- Impact of κ_{nanowire} on self-heating -



Effective self-heating in nanowire (low κ)

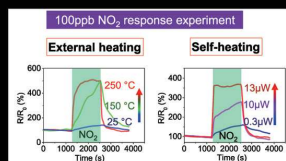
- Extract κ_{nanowire} by 3ω method -



SnO₂ bulk [100] 55 W/mK
 SnO₂ [100] nanowire ~ 2 W/mK

Reduced κ_{nanowire} (increased phonon scattering)

- Model verification -

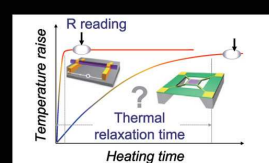


13 μW --- ~250 °C

► Low κ_{nanowire} facilitates localization of Joule heat within nanowire

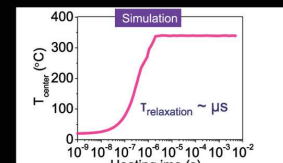
Reduce Energy consumption

- Concept: pulse measurement -

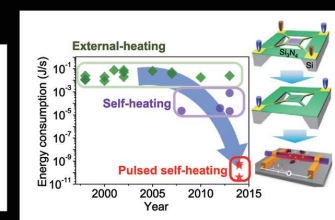


Read R just after temperature raising

- $\tau_{\text{relaxation}}$ -

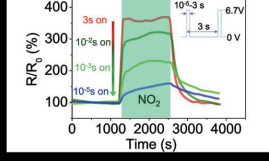


- Summary of energy consumption -



5 orders reduction of energy consumption!

- Pulsed self-heating -

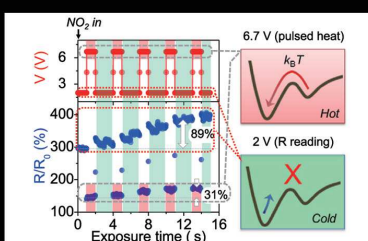


Showing response by a pulse (10 μs) heat

► Short $\tau_{\text{relaxation}}$ enables a drastic reduction of energy by pulsed mode

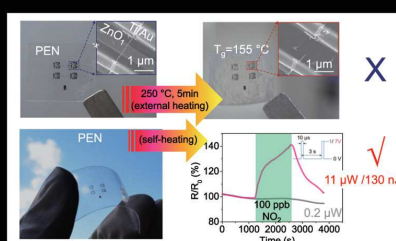
Further merits provided by pulsed self-heating

- Higher sensitivity -



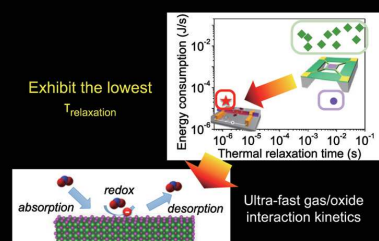
► Manage heat & R reading --- high S

- Good compatibility with flexible substrate -



► Overcome the bottleneck of working temperature

- Ultra-fast temperature modulation -



► New feature to distinguish gas molecules